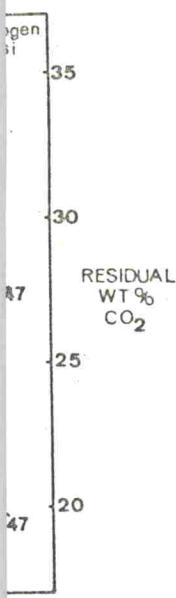


KINETICS IN CARBONATE-HYDROGEN SYSTEMS

TTI



% CH₄ in the solid for the 8, and 12 hour experiments.

Experiments were run at a range of reaction gases for the dolomite-hydrogen system.

are considerably more limited in evaluation of the rate of reaction time pair at 620°C. The latter, shows wide scatter, divergence from linearity is seen in a plot of the six temperature kinetic data into physically limited data available.

mesh siderite fragments containing calcite-hydrogen and studied with only four experiments up to 605°C and 2000 torr helium was used in

TABLE 4. MASS SPECTROGRAPHIC ANALYSES OF REACTION GASES above AND WET-CHEMICAL ANALYSES of RESIDUAL CO₂ IN UNREACTED DOLOMITE: DOLomite-HYDROGEN SYSTEM

Experiment No.	39	41	42	43	44	45	46	47	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64
Temperature (°C)	520	525	550	620	735	620	620	735	735	735	735	735	735	735	735	735	735	735	735	735	735	735	735	735
Pressure (psi)	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000	5,000
Wt. % CO ₂ remaining in solid	N.A. ^d	44.3	40.2	25.6	10.1	25.4	30.7	19.4	N.A.															
Duration of run (hours)	4	4	4	4	4	8	2	12	4	4	4	4	4	4	4	4	4	6	6	7	7	4	2	4
Mole % CH ₄ in gas	0.01	0.02	0.04	1.20	2.34	1.82	1.53	2.05	2.00	1.88	1.77	1.80	1.77	1.77	1.77	1.77	1.77	1.77	1.77	1.77	1.77	1.77	1.77	1.77
Mole % C ₂ H ₆ in gas	—	—	0.01	0.02	—	0.04	0.05	0.04	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
Mole % CO ₂ in gas	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
Mole % CO in gas	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
Mole % H ₂ in gas	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—
Mole % He in gas	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—	—

a. Analysis calculated on water free basis.

b. Starting hydrogen impurities given as follows: less than 5 ppm O₂, less than 1/2 ppm N₂, less than 1/2 ppm CO₂, less than 1/2 ppm CO, Dew point = 100°F.

c. Detection limit 0.01%.

d. Not analyzed.